



5.00 credits

30.0 h + 30.0 h

Q1

Teacher(s)	Demoustier Sophie ;Fustin Charles-André ;
Language :	French
Place of the course	Louvain-la-Neuve
Prerequisites	LEPL1301 (Chemistry 1) or a similar course
Main themes	The course is divided in two parts. The first part deals with 'generalities' required for a good understanding of the reactivity in organic chemistry, namely the reasons why organic compounds do or do not react in given conditions. The second part describes the chemical behavior of the main organic compounds, illustrating the relationships between the structure of a given functional group and its reactivity. The lessons will be frequently illustrated with examples from other disciplines such as materials science and life sciences.
Learning outcomes	<p>At the end of this learning unit, the student is able to :</p> <p>Contribution of the course to the program objectives</p> <p>Regarding the learning outcomes of the program of Bachelor in Engineering Sciences, this course contributes to the development and the acquisition of the following learning outcomes:</p> <p>LO 1.1 : Apply concepts, laws, reasoning to disciplinary reduced problems.</p> <p>Specific learning outcomes of the course</p> <p>At the end of the course, the student will be able to:</p> <p>1</p> <ul style="list-style-type: none"> • distinguish the different types of isomers (structure and geometrical isomers and stereoisomers) ; • recognize the different types of reactants (nucleophiles, electrophiles, radicals, acids and bases) ; • describe the electrons migration within an organic molecule (inductive and conjugation effects) as well as during a chemical reaction between two given compounds; • recognize and represent the main functional groups of organic compounds ; • establish relations between molecular and spatial structures of organic molecules and some properties, in particular, their reactivity ; • predict and explain the expected result for the main types of organic reactions, including their mechanism.
Evaluation methods	Students are evaluated through a final written exam.
Teaching methods	The course is based on lectures and exercises-based learning
Content	<p>Part 1</p> <ol style="list-style-type: none"> 1. Structure, chemical bonds and geometry of organic molecules 2. Isomerism 3. Reactivity in organic chemistry (energy diagrams, intermediates, types of reactants, electronic effects) <p>Part 2</p> <ol style="list-style-type: none"> 4. Alkanes , alkenes and alkynes 5. Alkyl halides 6. Aromatic compounds 7. Alcohols, thiols, ethers and epoxydes 8. Aldehydes and ketones 9. Carboxylic acids and their derivatives 10. Amines and their derivatives
Inline resources	http://moodleucl.uclouvain.be/course/view.php?id=8644
Bibliography	<ul style="list-style-type: none"> • Les slides présentées au cours et les énoncés des exercices sont disponibles sur Moodle. <p>Ouvrages de référence recommandés mais non obligatoires:</p> <ul style="list-style-type: none"> • L. Craine, D. Hart, C. Hadad, Chimie Organique 1 et 2, Dunod, 2008 • D. Klein, Organic Chemistry, Wiley, 2011

Faculty or entity in charge	FYKI
-----------------------------	------

Programmes containing this learning unit (UE)				
Program title	Acronym	Credits	Prerequisite	Learning outcomes
Minor in Applied Chemistry and Physics	MINOFYKI	5		
Specialization track in applied Chemistry and Physics	FILFYKI	5		
Mineure Polytechnique	MINPOLY	5		