UCLouvain

2022

## General and Organic Chemistry



## This learning unit is not open to incoming exchange students!

Teacher(s)	Glinel Karine ;Luis Alconero Patricia ;Norberg Valérie ;Pouyez Jenny ;				
Language :	French				
Place of the course	Charleroi				
Prerequisites	The student must have the basic notions of chemistry as taught in science courses in general secondary education namely: basic concepts such as understanding the notion of mole, atomic mass, molar and relative molecular Avogadro's number, density, concentration and specific gravity. He must also know the symbols of the chemic elements, the nomenclature used in general chemistry and the main organic chemical functions.				
Main themes	The first part of the course will deal with elements of general chemistry and thermodynamics for the understanding of chemical structures, interactions and reactivity. In a second part, the major classes of organic compounds will be presented as well as the main chemical reactions useful for understanding certain biochemical or biological phenomena.				
Learning outcomes	At the end of this learning unit, the student is able to :				
	S1.G5At the end of this teaching unit, the student is able to:				
	S5.2, - to master the molecular structures of the main organic compounds, their interactions and their S5.3, transformations by chemical reactions:				
	S5.4, - to master the concepts making it possible to quantify a chemical reaction both from the point of view of S5.5, the material balance and the energy balance;				
	S6.1, S6.2, - apply computer science skills to general and organic chemistry such as the use of software for the simulation of chemical reaction monitoring.				
Evaluation methods	Students are evaluated on the basis of:				
L valuation methods	1) their work during the quarter and				
	2) a final exam.				
	The final grade is based on:				
	- a written or oral exam at the end of the semester which counts for 75% of the final grade				
	- two laboratory reports which count for 25% of the final grade				
	The mark related to the laboratories is acquired for all sessions of the academic year, by virtue of art. 78 of th RGEE				
	A written quiz during the quarter may allow to be exempted from a part of the questions of the final exam if the mark obtained for this quiz is equal to or higher than 13/20. The parts of the course covered by this exemption will be communicated to the concerned students after the quiz.				
	Participation in practical work sessions (laboratories) organized during the year is required.				
Teaching methods	The course is based on :				
g even	- Lectures given in person and/or by video-conference				
	- video clips describing some concepts covered during the course				
	- exercise sessions				
	- practical work sessions (laboratories).				
Content	The first part of the course deals with the elements of general chemistry and thermodynamics for the understandin of chemical structures, interactions and reactivity. In a second part, the main classes of organic compounds will b presented as well as the main chemical reactions useful to understand some biochemical or biological phenomena				
Inline resources	The slides used during the lectures, the video capsules as well as the statements of the exercises and labs are made available via the Moodle website of the course.				
Bibliography	Les ouvrages suivants sont recommandés :				
Dibilography	1- Chimie générale, une approche moléculaire, 2e édition   (Français) Broché – 27 avril 2018				
	de Julie Vézina (Adapté par), Nivaldo J. Tro (Avec la contribution de), Jean-Marie Gagnon (Avec la contribution de				
	2- Chimie organique simple et intuitive - David Klein – Traduction française P. Depovere – Edition Deboeck.				

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Other infos	The student taking this course must have basic notions of chemistry such as the notion of mole, atomic, molar and relative molecular mass, Avogadro number, density, concentration and density. He/she must also know the symbols of chemical elements, the nomenclature used in general chemistry and the main organic chemical functions.
Faculty or entity in charge	SINC

Programmes containing this learning unit (UE)							
Program title	Acronym	Credits	Prerequisite	Learning outcomes			
Bachelor in Computer Science	SINC1BA	5		•			