

Due to the COVID-19 crisis, the information below is subject to change, in particular that concerning the teaching mode (presential, distance or in a comodal or hybrid format).


9 credits

60.0 h + 30.0 h

Q1

Teacher(s)	Rider Mark (coordinator) ;Vlad Alexandru ;
Language :	French
Place of the course	Bruxelles Woluwe
Main themes	The formation is oriented towards problems solving. Formal lessons are given and activities in small groups are organized where numerical chemistry problems are worked out. The topics covered are atomic, ionic and molecular properties, conservation of matter, gas properties, reactivity, thermodynamics, equilibria in aqueous solution and kinetics.
Aims	<p>The aim of the course is to give a basic knowledge of general chemistry to students oriented towards life sciences. With these lessons, the students should acquire a sound idea of what atoms and molecules are and how they behave. They should be able to use in a proper fashion the basic notions of molecular structure, reactivity, thermodynamics and kinetics. At the end of half an academic year, typical numerical problems of a first year college chemistry course have to be mastered.</p> <p>1</p> <p>-----</p> <p><i>The contribution of this Teaching Unit to the development and command of the skills and learning outcomes of the programme(s) can be accessed at the end of this sheet, in the section entitled "Programmes/courses offering this Teaching Unit".</i></p>
Evaluation methods	<p><b>Due to the COVID-19 crisis, the information in this section is particularly likely to change.</b></p> <p>Written exam involving solving problems but also including explanations of theoretical aspects of the course (± 12 questions).</p> <p>There are no negative points or ponderation according to the questions and course content. However, when a student has a mark between 9/20 et 10/20 after correction, the lecturers re-examine all the points to decide if the mark should be rounded up or down. If the answers are considered to be insufficient (for example failure of more than half of the questions), the final mark will be rounded down to 9/20.</p>
Teaching methods	<p><b>Due to the COVID-19 crisis, the information in this section is particularly likely to change.</b></p> <p>Teaching is by formal lectures. There are a series of practicals (TP) and demonstrations (TD) that accompany the teaching of this course.</p> <p>The teaching will be conducted face-to-face or at distance exclusively or partially according to health restrictions.</p>
Content	<p>This course of general and inorganic chemistry forms part of the curriculum for first year pharmacy (FARM) and biomedical science (SBIM) students.</p> <p>Content:</p> <ol style="list-style-type: none"> <li>1. Introduction</li> <li>2. Atoms : description and properties</li> <li>3. Chemical bonding : ionic, covalent and metallic (nature, stability)</li> <li>4. Chemical thermodynamics</li> <li>5. States of matter</li> <li>6. Chemical equilibria</li> <li>7. Acids and bases</li> <li>8. Solubility</li> <li>9. Electrochemistry</li> <li>10. Chemical kinetics</li> </ol>
Inline resources	There is no formal syllabus ! PDF versions of slides presented in the course, which cover the subject in a comprehensive way, will be made available on MoodleUCL ( <a href="https://moodleucl.uclouvain.be/">https://moodleucl.uclouvain.be/</a> ). In addition, a tablet will be used to explain certain aspects of the course. The "Tablet" PDF versions of the PowerPoint files will also be made available to students via MoodleUCL.
Bibliography	<p>Livres de référence :</p> <ul style="list-style-type: none"> <li>• Atkins, Jones, Principes de Chimie (de boek, 2ème/3ème édition)</li> <li>• Chimie des Solutions, Kotz, Treichel Jr, de boek/Beauchemin</li> <li>• Ayadim, Habib-Jiwan, Chimie Générale Edition : UCL press Universitaires de Louvain-DUC- 2013.</li> <li>• Voir aussi <a href="http://www.deboek.com">www.deboek.com</a> et <a href="http://www.lachimie.org">www.lachimie.org</a> (site très utile pour travailler son cours).</li> </ul>

Other infos	<p>The participation in the series of demonstrations (TD) and exercises is indispensable and strongly recommended. Participation in the practicals (TP) is obligatory and unjustified absences could lead to a penalty (0/20 for the exam).                  "En cas d'absences répétées même justifiées, l'enseignant peut proposer au jury de s'opposer à l'inscription à l'examen relatif à l'UE en respect de l'article 72 du RGEE*"</p>
Faculty or entity in charge	FASB

<b>Programmes containing this learning unit (UE)</b>				
Program title	Acronym	Credits	Prerequisite	Aims
Bachelor in Pharmacy	<a href="#">FARM1BA</a>	9		
Bachelor in Biomedicine	<a href="#">SBIM1BA</a>	9		