UCLouvain Imapr1230
2018 Organic chemistry
5 credits 45.0 h + 15.0 h Q2

Teacher(s)	Demoustier Sophie ;Elias Benjamin ;Fustin Charles-André (compensates Elias Benjamin) ;Mignon Denis ;					
Language :	French					
Place of the course	Louvain-la-Neuve					
Prerequisites	FSAB 1301 (Chemistry 1) or a similar course The prerequisite(s) for this Teaching Unit (Unité d'enseignement – UE) for the programmes/courses that offer this Teaching Unare specified at the end of this sheet.					
Main themes	The course is divided in three parts. The first part deals with 'generalities' required for a good understanding of the reactivity in organic chemistry, namely the reasons why organic compounds do or do not react in given conditions. The second part describes the chemical behavior of the main organic compounds, illustrating the relationships between the structure of a given functional group and its reactivity. The lessons will be frequently illustrated with examples from other disciplines such as materials science and life sciences. The third part consists in an introduction to the main separation techniques used in the chemical industry and to the oil refining industry in particular. It addresses the thermodynamics of phase equilibria (liquid-liquid and liquid-vapor) and describes some practical applications of these equilibria, such as liquid-liquid extraction or fractional distillation.					
Aims	Contribution of the course to the program objectives Regarding the learning outcomes of the program of Bachelor in Engineering Sciences, this course contributes to the development and the acquisition of the following learning outcomes: LO 1.1 : Apply concepts, laws, reasoning to disciplinary reduced problems. Specific learning outcomes of the course At the end of the course, the student will be able to: distinguish the different types of isomers (structure and geometrical isomers and stereoisomers); recognize the different types of reactants (nucleophiles, electrophiles, radicals, acids and bases); describe the electrons migration within an organic molecule (inductive and conjugation effects) as well as during a chemical reaction between two given compounds; establish relations between molecular and spatial structures of organic molecules and some properties, in particular, their reactivity; predict and explain the expected result for the main types of organic reactions, including their mechanism; explain how the chemical process industry takes advantage of the various equation of state- or activity coefficients-based models to compute thermodynamic properties of liquid-vapor or liquid-liquid mixtures at equilibrium; explain the operation principles of a distillation columns for binary mixtures, of a crude oil distillation column or of a liquid-liquid extraction; explain the main units composing an oil refinery.					
Evaluation methods	Students are evaluated through a final written examination					
Teaching methods	The course is based on lectures and exercises-based learning					
Content	Part 1 1. Structure, chemical bonds and geometry of organic molecules 2. Isomerism 3. Reactivity in organic chemistry (energy diagrams, intermediates, types of reactants, electronic effects) Part 2 4. Alkanes, alkenes and alkynes 5. Alkyl halides 6. Aromatic compounds					

Université catholique de Louvain - Organic chemistry - en-cours-2018-Imapr1230

1	7. Alcohols, thiols, ethers and epoxydes				
	8. Aldehydes and ketones				
	9. Carboxylic acids and their derivatives				
	10. Amines and their derivatives				
	Part 3				
	11. Phase equilibria : real systems with one or several constituents				
	12. Practical applications of phase equilibria				
	13. Introduction to the refining industry				
Inline resources	http://moodleucl.uclouvain.be/course/view.php?id=8644				
Bibliography	• Les slides présentées au cours et les énoncés des exercices sont disponibles sur Moodle.				
1	Ouvrages de référence recommandés:				
	 L. Craine, D. Hart, C. Hadad, Chimie Organique 1 et 2, Dunod, 2008 D. Klein, Organic Chemistry, Wiley, 2011 				
Faculty or entity in	FYKI				
charge					

Programmes containing this learning unit (UE)							
Program title	Acronym	Credits	Prerequisite	Aims			
Bachelor in Engineering	FSA1BA	5	LEPL1301	٩			
Minor in Engineering Sciences : Applied Chemistry and Physics	LFYKI100I	5		•			