UCLouvain

Ichm2181

2017

Homogeneous and heterogeneous catalysis

Teacher(s)	Gaigneaux Eric ;Riant Olivier coordinator ;			
Language :	French			
Place of the course	Louvain-la-Neuve			
Main themes	Generalities on catalysis. Recall of the basic principle of catalysis. Homogeneous, heterogeneous catalysis and heterogenisation of homogeneous catalysis. Description and comparison of elementary processes in these two categories of catalysis. Part I: homogeneous catalysis 1. Basic principles of homogeneous catalysis Recall of basics in organometallic chemistry (16-18 electron rule, classification of ligands, donation and retrodonation) Elementary reactions: ligand substitution, oxidative addition, reductive elimination, insertion, oxidative coupling Ancillary ligands: mono and di-phosphines case, cone angles, diaminocarbenes ligands Principle of catalysis by phase transfer. 2. Some examples of important industrial processes using homogeneous catalysis. Examples: alkene hydrocyanation (nylon channel), alkene hydroformylation, acid acetic synthesis, oxidation processes, biphasic processes. Part II: heterogeneous catalysis 1. Basic principles of heterogeneous catalysis. 2. Review of principal kinetic models of heterogeneous catalysis (Langmuir-Hinshelwood, Eley-Rideal, Mars -van Krevelen) 3. Illustrations in the following areas: petrochemistry (hydrotreatment, catalytic craking, oxidation and ammoxidation of propene, epoxidation of ethylene, oxidation of butane in maleic anhydride, use of zeolites,), environment (DeNOx, exhaust gases) and basic inorganic chemistry synthesis (ammonia, sulphuric acid,).			
Aims	This teaching aims (1) at providing future chemists a global and unified vision of catalysis, describing and comparing elementary processes in homogeneous and heterogeneous catalysis, and (2) at illustrating the importance of the two categories of catalysis at the industrial level. The contribution of this Teaching Unit to the development and command of the skills and learning outcomes of the programme(s) can be accessed at the end of this sheet, in the section entitled "Programmes/courses offering this Teaching Unit".			
Other infos	The course could be partly or totally delivered by an invited lecturer.			
Faculty or entity in charge	СНІМ			

Programmes containing this learning unit (UE)					
Program title	Acronym	Credits	Prerequisite	Aims	
Master [120] in Chemistry	CHIM2M	3		Q.	
Master [60] in Chemistry	CHIM2M1	3		0	