### Physical Chemistry for Metals and Ceramics

**2016-2017**

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<tr>
<th>Credits</th>
<th>Hours</th>
<th>Semester</th>
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<td>5.0</td>
<td>30.0 + 30.0</td>
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**Teacher(s):** Jacques Pascal

**Language:** Anglais

**Place of the course:** Louvain-la-Neuve


**Main themes:**
Three main themes will be considered: non metallic inorganic solids (ceramics and mineral glasses), physical metallurgy, main processing of the principal industrial metals and alloys (steel, aluminum, ...).

**Aims:**

Contribution of the course to the program objectives

With respect to the general objectives of the KIMA program, the present course contributes to the development of the following learning outcomes:

- **AA1 Scientific and technical knowledge** (AA1.1, AA1.3)
- **AA2 Engineering competences** (AA2.1, 2.2)
- **AA3 R&D competences** (AA3.1)

Specific learning outcomes of the course

At the end of the course, the students should be able to:

- **AA1.1** to describe the solid state phase transformations in binary metallic systems;
- **AA1.1** to give the general characteristics of the most common engineering metallic materials;
- **AA1.1** to describe the equilibria in ternary systems;
- **AA1.1** to describe the structure of ceramics and inorganic glasses, as well as the point defects;
- **AA1.1** to give the mechanisms of mass and charge transport in ceramics;
- **AA1.3, AA2.1** to analyse the influence of the production processes used for inorganics materials;
- **AA1.3, AA2.1** to choose the mostly adapted category of inorganic materials for a specific application based on requested performances and economical concerns.

### Evaluation methods:

The students are evaluated individually with a written and oral exam based on the objectives described above.

The written exam will concern the scientific and technical knowledge seen during the lectures as well as the resolution of an exercise related to the physical chemistry of metals and ceramics.

A specific exam will deal with the practicals. It consists in the recognition of the microstructure of specific samples and solving specific problems.

### Teaching methods:

The course is organised around 12/13 lectures, 4 exercise sessions and 6 laboratories.

### Content:

- **Phase equilibria:** reminder of binary systems; ternary equilibria;
- **Diffusion in solids:**
- **Ceramics and glasses:** crystal structure, amorphous state, mineral glasses, crystalline defects, transport of mass and electrical conductivity, phase equilibria in processes;
- **Physical metallurgy:** types of transformations, kinetics of transformations, TTT diagrams, application to different metals and alloys (steels, aluminum, ...);
- **Steelemaking processes**;
- **Extrative metallurgy of aluminum**.

### Bibliography:

A syllabus is available via SICI. The teacher will also deposit actualised versions of the syllabus on the web site of the course on iCampus.


### Other info:

It is highly recommended to have knowledge in chemistry and physical chemistry at the bachelor level. Particularly, the courses LMAPR 1805 : Introduction à la science des matériaux, LMAPR 1310 : Thermodynamique * équilibres de phase et LMAPR 1231 : Procédés de chimie inorganique.
| Faculty or entity in charge: | FYKI |
### Programmes / formations proposant cette unité d'enseignement (UE)

<table>
<thead>
<tr>
<th>Intitulé du programme</th>
<th>Sigle</th>
<th>Credits</th>
<th>Prerequis</th>
<th>Acquis d'apprentissage</th>
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<tr>
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<tr>
<td>Master [120] in Chemical and Materials Engineering</td>
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<tr>
<td>Master [120] in Chemistry and Bioindustries</td>
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