

4.0 credits	30.0 h + 15.0 h	1q
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Teacher(s) :	Bartosiewicz Yann ;
Language :	Français
Place of the course	Louvain-la-Neuve
Inline resources:	Icampus
Prerequisites :	Physics and thermodynamics from previous years
Main themes :	<ol style="list-style-type: none"> <li>1. The two principles of thermodynamics and their consequences</li> <li>2. The thermodynamics of gas</li> <li>3. The thermodynamics of vapors</li> <li>4. Engine cycles</li> <li>5. Operating cycles</li> </ol>
Aims :	<p>Contribution of the course to the reference learning outcomes B1.1 ;1.5 ;2.1 ;4.2</p> <p>b. Course-specific learning outcomes Through fundamentals concepts taught in class plus practices, the student should be able to:</p> <ol style="list-style-type: none"> <li>1. Perform an energy balance and evaluate the losses together with the energy deterioration across a thermodynamic transformation involving heat and mechanical work exchanges;</li> <li>2. To compute and propose a simple model of an engine thermal cycle involving a perfect gas or a vapor, e.g. Rankine cycles, gas turbine cycles, internal combustion engine cycles, or cogeneration;</li> <li>3. To compute and propose a simple model of a refrigeration cycle, e.g. compressions machines, heat pumps or absorption;</li> </ol> <p><i>The contribution of this Teaching Unit to the development and command of the skills and learning outcomes of the programme(s) can be accessed at the end of this sheet, in the section entitled "Programmes/courses offering this Teaching Unit".</i></p>
Evaluation methods :	Closed book exam made of one theoretical and one application (exercices) parts. A personal form for mathematical expressions/formula is allowed.
Teaching methods :	Lectures are presented by using slides, particular demonstrations and details are given at the board as well as examples.
Content :	<ol style="list-style-type: none"> <li>1. The two principles of thermodynamics and the related variables (internal energy, enthalpy and entropy), the work and heat concepts, the fundamental equations of shaft work, opened/closed transformations, T,s and h,s diagrams;</li> <li>2. Perfect gas, properties and mixture of perfect gas, isentropic and polytropic transformations, models of compression/expansion;</li> <li>3. Vaporization, triple point, critical point, vapor phase diagram, heat of vaporization, saturation conditions, overheat, properties of vapors;</li> <li>4. Internal combustion engine, vapor installations, cogeneration;</li> <li>5. Carnot cycle, compression cycles, absorption cycles, heat pumps;</li> </ol>
Bibliography :	Syllabus « THERMODYNAMIQUE », BIR1311, J. Martin et P. Wauters, ed. 2012 « Fundamentals of engineering thermodynamics », M. J. Moran and H. N. Shapiro
Cycle and year of study :	<a href="#">&gt; Master [120] in Environmental Bioengineering</a> <a href="#">&gt; Bachelor in Bioengineering</a>
Faculty or entity in charge:	AGRO